Specificity in the hydrolysis of N-acyl-L-phenylalanine 4-nitroanilides by chymotrypsin

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Summary. The affinity of N-acyl-L-phenylalanine 4-nitroanilides for chymotrypsin is enhanced as the hydrophobicity of non-amino acid residues in the P_2 -position of the substrates increases, whereas k_{cat} remains nearly constant. On the other hand, if alanine or leucine is in the P_2 -position k_{cat} increases with decreasing K_M .

The binding of specific peptide substrates to chymotrypsin occurs by hydrophobic interactions in the so-called tosyl pocket² as well as by subsite interactions ³⁻⁷. The inactivation of chymotrypsin by peptidyl chloromethyl ketones in concert with X-ray crystallographic studies^{8,9} reveals that there are, besides hydrogen bonds, additional hydrophobic interactions of the amino acid residue in the P₂-position (nomenclature according to Schechter and Berger¹⁰ of the inhibitor with the subsite S₂ of the enzyme. The P₂-S₂ contact is also documented by the enzymic hydrolysis of oligopeptide amides^{6,11} and denatured proteins¹².

To get more information on the influence of hydrophobic N-acyl residues in chymotrypsin substrates, we have studied the hydrolysis of differently substituted L-phenylalanine 4-nitroanilides by chymotrypsin. Glt-, Glt- ε Ahx-, Glt- $(\varepsilon$ Ahx)₂-, Glt-Ala-, and Glt-Leu- were used as N-acyl residues

Material and methods. Bovine a-chymotrypsin (chymotrypsin A_4) was purchased from Boehringer, Mannheim. The content of active sites (89%) in the protein was determined by titration with 4-nitrophenyl acetate¹³. The protein concentration was calculated by the use of $A_{280}^{1\%} = 20.0$.

Preparation of the substrates. Glt-Phe-Nan was synthesized according to Erlanger et al. ¹⁴. The preparation of Glt-εAhx-Phe-Nan and Glt-εAhx)-Phe-Nan has been described earlier ¹⁵. Glt-Ala-Phe-Nan, m.p. 180–183 °C, $[a]_D^{22}$ –10.0° (c=1, methanol) and Glt-Leu-Phe-Nan, m.p. 178–179 °C, $[a]_D^{22}$ –9.0° (c=1, methanol) were synthesized from Boc-Ala-Phe-Nan, m.p. 167–170 °C, $[a]_D^{21}$ –16.4° (c=1, ethyl acetate) and Boc-Leu-Phe-Nan, m.p. 98–99 °C, $[a]_D^{21}$ –13.7° (c=1, ethyl acetate) by removal of the Boc-groups and subsequent acylation according to the procedure used for Glt-Phe-Nan.

All elemental analyses were within acceptable limits. TLC also indicated homogeneity [chloroform-methanol (7:1), n-butanol-acetic acid-water (4:1:1), isobutanol-formic acid-water (75:13.5:11.5)]. The degree of enzymic hydrolysis of all substrates was > 97%.

Kinetic measurements. All measurements were carried out in 0.05 M Tris HCl buffer, 0.05 M CaCl₂, 1% (v/v) dimethylformamide, pH 7.8, at 25 °C using a Perkin-Elmer M 356 or a Beckman DK-2A spectrophotometer. Enzymic hydrolysis was followed by the increase in absorption of the liberated 4-nitroaniline at 405 nm (ε = 9800 M⁻¹ cm⁻¹) and 390 nm (ε = 11,300 M⁻¹ cm⁻¹). K_M and k_{cat} were calculated from the slopes and intercepts of the Lineweaver-Burk and Eadie plots. 5-7 runs were performed for each determination of K_M and k_{cat}.

The direct determination of the 2nd-order rate constant k_{cat}/K_M was carried out under pseudo 1st-order conditions. The conditions were fulfilled by $[S] \le 0.1~K_M$ and $[E] \le 0.01~K_M$. k_{cat}/K_M was calculated from the pseudo 1st-order rate constant k_{cat} . $[E]/K_M$.

All data were obtained by 3 independent measurements. The standard deviations of the constants were 7-12%.

Results and discussion. The hydrolysis of ester and amide substrates by chymotrypsin can be described as follows⁴:

$$E+S \xrightarrow{K_s} ES \xrightarrow{k_2} EA+P_1 \xrightarrow{k_3} E+P_2$$

For amides and anilides, acylation is the rate-limiting step, i.e. $k_{cat} = k_2$ and $K_M = K_s^{\ 4}$.

Kinetic parameters listed in the table show that there are significant differences in the hydrolysis between the substrates containing either Ac, Glt, and ε Ahx or L-amino acid residues (Ala, Leu) in the P_2 -position. In the series of Ac, Glt-, Glt- ε Ahx-, and Glt-(ε Ahx)₂-Phe-Nan K_M decreases as the hydrophobicity of the N-acyl residues increases, whereas k_{cat} remains nearly constant. In contrast to the substrates mentioned, k_{cat} increases with decreasing K_M values in the hydrolysis of Glt-Ala- and Glt-Leu-Phe-Nan. The K_M values of Glt-Ala- and Glt-Leu-Phe-Nan are higher than the corresponding values of the substrates without an α -amino acid residue in P_2 . On the other hand, 19-fold and 64-fold k_{cat} values were found for Glt-Ala- and Glt-Leu-Phe-Nan compared with Glt- ε Ahx-Phe-Nan.

The high k_{cat} values of Glt-Ala- and Glt-Leu-Phe-Nan can be explained by specific enzyme-substrate interactions, which bring the scissile bond into a favourable position to the catalytic residues of the enzyme. X-ray studies of Aγ-chymotrypsin modified by N-acetyl tripeptide chloromethyl ketones indicate 3 hydrogen bonds between enzyme and inhibitor⁸. 2 of these hydrogen bonds (between NH of P₁ and CO of Ser-214; CO of P₃ and NH of Gly-216) can be formed in the binding of Glt-Ala (Leu)-Phe-Nan. The hydrophobic interactions between P₂ and S₂ were confirmed by the inactivation rate of chymotrypsin by Ac-Leu-Phe-CH₂Cl that was 2.9 times faster than that by Ac-Ala-Phe-CH₂Cl⁹. Similarly, we found that the ratio of the k_{cat} values of Glt-Leu-Phe-Nan and Glt-Ala-Phe-Nan amounts to 3.3.

In the chymotrypsin-catalyzed hydrolysis of peptide amides, which contain an aliphatic a-amino acid residue in P_2 , k_{cat} increases with decreasing $K_M^{6,11}$. Despite the low K_M values the k_{cat} values of Glt- $(\varepsilon Ahx)_{1,2}$ -Phe-Nan are much lower than the k_{cat} values of Glt-Ala (Leu)-Phe-Nan. This behaviour of Glt- $(\varepsilon Ahx)_{1,2}$ -Phe-Nan in the enzymic hydrolysis can be interpreted as follows. The low K_M values should

Kinetic parameters for the α -chymotrypsin-catalyzed hydrolysis of N-acyl-L-phenylalanine 4-nitroanilides

Substrate	[S] (mM)	K _M (mM)	k _{cat} (sec ⁻¹)	$\frac{k_{cat}/K_M}{(M^{-1} \cdot sec^{-1})}$
Ac-Phe-Nan*		0.40	0.01	(25)
Glt-Phe-Nan	0.10 - 1.0	0.25	0.021	84 (96)
Glt-Phe-Nan**	_	0.28	0.013	46 `´
Glt-&Ahx-Phe-Nan	0.04 - 0.8	0.056	0.02	350
Glt-(&Ahx)2-Phe-Nan	0.01-0.5	0.016	0.019	1180
Glt-Ala-Phe-Nan	0.10-2.0	1.00	0.38	380 (418)
Glt-Leu-Phe-Nan	0.08 - 0.8	0.50	1.28	2560 (2680)

Measurements were carried out in 0.05 M Tris · HCl buffer, 0.05 M CaCl₂, pH 7.8, containing 1% (v/v) dimethylformamide, at 25 °C. The enzyme concentration varied in the range from 1 to 10 μ M. The k_{cat}/K_M values in parentheses were determined directly under pseudo 1st-order conditions. For details see methods section. * The values were taken from Petkov et al. ¹⁶ and ** from Erlanger et al. ¹⁴

result from hydrophobic interactions of the Glt- $(\varepsilon Ahx)_{1,2}$ -residues with S_2 and/or other apolar sites on the protein surface. However, this contact does not lead to an acceleration of catalysis due to the failure of the hydrogen bond between CO of P_3 and NH of Gly-216. In the series of Ac-, Glt-, Glt- $(\varepsilon Ahx)_{1,2}$ -Phe-Nan K_M decreases, whereas k_{cat} is nearly constant. Therefore, non-productive binding of the N-acyl residues should not be important, since such interactions lower both K_M and $k_{cat}^{17,18}$. In addition, non-productive binding of the aniline moiety could not be substantiated.

- Abbreviations: Ac-, acetyl-; εAhx-, 6-aminohexanoyl-; Boc-, tert.-butyloxycarbonyl-; Glt-, glutaryl-; Nan-, 4-nitroanilide.
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The high affinity of $Glt-(\varepsilon Ahx)_{1,2}$ -Phe-Nan to the enzyme is reflected in the reaction rate under pseudo 1st-order conditions ($[S] \ll K_M$). Thus, the k_{cat}/K_M value of $Glt-(\varepsilon Ahx)_2$ -Phe-Nan is 3 times higher than that of Glt-Ala-Phe-Nan. In contrast, the k_{cat} values of Glt-Ala-Phe-Nan is 20 times higher than the k_{cat} value of $Glt-(\varepsilon Ahx)_2$ -Phe-Nan.

higher than the k_{cat} value of Glt- $(\varepsilon Ahx)_2$ -Phe-Nan. The k_{cat} and k_{cat} / K_M values in the table show that Glt-Leu-Phe-Nan is more specific than the other substrates studied. Glt-Leu-Phe-Nan should also be of practical interest as a chromophoric chymotrypsin substrate because it is hydrolyzed much faster than the frequently used Glt-Phe-Nan.

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C-banding pattern on the chromosomes of the Japanese house shrew, Suncus murinus riukiuanus, and its implication

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Summary. The C-band on the chromosomes of the Japanese house shrew, Suncus murinus riukiuanus (Insectivora), was studied. Various types of C-banding pattern were found in the genome of this subspecies. Such banding patterns could be useful for an understanding of autosome and sex-chromosome polymorphisms within S. murinus.

Karyotypes of the house shrew, Suncus murinus, occurring in various areas of Asia have already been studied, and autosomal and sex chromosomal polymorphisms have been found in this species³⁻⁷, but information on the C-band of the house shrew has as yet only been obtained for the Indian taxon⁷. Nevertheless, C-band staining is a good tool for examining polymorphisms. In this paper, we report the C-banding pattern for the genome of the Japanese house shrew, S.m.riukiuanus, and also present its conventional data again here, because the karyotype figure has not been published in detail.

Materials and methods. 3 male specimens collected from Naha, Okinawa Is., Prefecture of Okinawa, were investigated karyologically. The C-band treatment followed the method of Sumner, and both conventional staining and C-band treatment were performed on the same preparation. For chromosomal classification the method of Patton was adopted.

Results. The C-banding patterns on the autosomes of S.m. riukiuanus are as follows. Within the M · SM-elements (row 1, figure 1), a large pair has a centromeric C-band, which is difficult to detect (see also figure 2A); a small pair and a medium-sized one have a distinct centromeric heterochromatic region; and another small pair exhibits terminal C-bands on its short arm, but does not have centromeric C-band. 2 pairs of ST-autosomes are C-band negative. Of the A-autosomes (row 2 and 3, figure 1), only 5

pairs having a short arm were C-band positive; the remaining 8 pairs seem to be devoid of demonstrable C-band material. 4 pairs of the former A-autosomes have both centromeric C-bands and heterochromatic short arms, but a medium-sized pair (placed in the 2nd position of the 2nd row) is distinguished by a lesser amount of C-banded material than is present in the other 3 pairs. On the other hand, the smallest acrocentric pair (5th in the 2nd row), has a terminal C-band only.

The C-banding pattern on the sex chromosomes (figures 1 and 2, B and C) differs markedly from that of the autosomes. The SM-X chromosome has a broad C-band covering the distal one-third of the long arm and a small terminal C-band on the short arm. The SM-Y chromosome

Lengths of the X and Y chromosomes relative to the female haploid set in the Japanese house shrew, Suncus murinus riukiuanus, calculated from 20 metaphase plates

	X Mean	SD	Y Mean	SD
Relative length	10.49 (5.81)	0.72 (0.35)	5.02	0.60

Values in parentheses show the relative length of only the euchromatic portion of the X chromosome.